Ref #	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
i i	12	(("6525230") or ("6472572") or ("6462243") or ("5998679") or ("5334777") or ("5243098")).PN.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	OFF	2006/02/11 18:35
L2	4094	((568/671) or (568/694) or (568/695) or (568/895) or (568/896) or (568/840) or (423/463) or (423/491) or (423/593) or (423/594) or (423/595) or (423/598) or (423/599)).CCLS.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	OFF	2006/02/11 18:37
13	444313	methane or alkane or ethane or propane or butane or pentane	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 18:38
L4	69349	metal\$4 near2 halide	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 18:38
L5	61	I2 and I3 and I4	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 18:39
L6	1654349	oxide	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 18:39
L7	37	I5 and I6	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:28
L8	37	waycuilis.in.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:29
L9	16	13 and 18	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:34

L10	177614	bromine	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:35
L11	4510		US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:35
L12	4046520	oxygen or air	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:36
L13	3879	l11 and l12	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:38
L14	44567	(reduced near2 metal) or (valence\$2 near2 state)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:39
L15	240	l13 and l14	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:40
L16	1415336	alcohol or ether	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:40
L17	216	l15 and l16	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:45
L18	4287620	water or steam or h2o!	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 19:45
L19	208	117 and 118	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 20:35

L20	16	regenerat\$6 near5 bromide near2 salt	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 20:43
L21	1	(second near2 bromide near2 salt) near10 bromine	US-PGPUB; USPAT; USOGR; EPO; JPO; DERWENT	OR	ON	2006/02/11 20:45
L22	363	(bromide near2 salt) near10 bromine	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 20:46
L23	118	I3 and I22	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 20:46
L24	116	l23 not l19	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 20:47
L25	79	I6 and I24	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON	2006/02/11 20:49
L26	3	l14 and l25	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	ON .	2006/02/11 20:48

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NEWS 12 JAN 17 IPC 8 in the WPI family of databases including WPIFV
NEWS 13 JAN 30 Saved answer limit increased
NEWS 14 JAN 31 Monthly current-awareness alert (SDI) frequency
                added to TULSA
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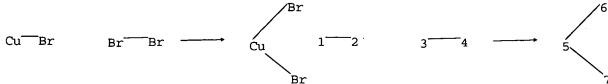
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chain nodes :
1 2 3 4 5 6 7
chain bonds :
1-2 3-4 5-6 5-7
exact bonds :
1-2 3-4 5-6 5-7

10/750,984

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS

fragments assigned product role:

containing 5

fragments assigned reactant/reagent role:

containing 1

containing 3

L1 STRUCTURE UPLOADED

=> que L1

L2 QUE L1

=> d

L2 HAS NO ANSWERS

L1

STR

Cu−Br Br → Cu

Structure attributes must be viewed using STN Express query preparation. L2 $$\operatorname{QUE}$$ L1

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BATCH **COMPLETE**

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PROJECTED ANSWERS: 0 TO

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FULL SEARCH INITIATED 21:05:09 FILE 'PS'

SCREENING COMPLETE - 0 REACTIONS TO VERIFY FROM 0 DOCUMENTS

100.0% DONE 0 VERIFIED 0 HIT RXNS 0 DOCS

SEARCH TIME: 00.00.01

0 L2 L3

=> s 12 ful

FULL SEARCH INITIATED 21:05:14 FILE 'CASREACT'

SCREENING COMPLETE - 14 REACTIONS TO VERIFY FROM 4 DOCUMENTS

100.0% DONE 14 VERIFIED 0 HIT RXNS 0 DOCS

SEARCH TIME: 00.00.01

FULL SEARCH INITIATED 21:05:15 FILE 'CHEMINFORMRX'

0 DOCUMENTS SCREENING COMPLETE - 0 REACTIONS TO VERIFY FROM

100.0% DONE 0 VERIFIED 0 HIT RXNS 0 DOCS

SEARCH TIME: 00.00.02

FULL SEARCH INITIATED 21:05:18 FILE 'DJSMONLINE'

SCREENING COMPLETE - 0 REACTIONS TO VERIFY FROM 0 DOCUMENTS

0 VERIFIED 0 HIT RXNS 100.0% DONE 0 DOCS

SEARCH TIME: 00.00.01

FULL SEARCH INITIATED 21:05:19 FILE 'PS'

SCREENING COMPLETE - 0 REACTIONS TO VERIFY FROM 0 DOCUMENTS

100.0% DONE 0 VERIFIED 0 HIT RXNS 0 DOCS

SEARCH TIME: 00.00.01

L4 0 L2

=> file caplus

COST IN U.S. DOLLARS

SINCE FILE TOTAL ENTRY SESSION 362.09 362.74 FULL ESTIMATED COST

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=> s bromine and bromide salt

50444 BROMINE

259443 BROMIDE

757227 SALT

560 BROMIDE SALT

(BROMIDE(W)SALT)

L5 37 BROMINE AND BROMIDE SALT

=> d 1-37 ti

- L5 ANSWER 1 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Catalytic conversion of alkanes into liquid oxygenates
- L5 ANSWER 2 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Enrichment of **Bromine** in Sea-Bittern with Recovery of Other Marine Chemicals
- L5 ANSWER 3 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Green technology in organic synthesis: Environmentally friendly waste free bromination of substituted acetophenones, benzocyclic ketones, and aromatic compounds
- L5 ANSWER 4 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI New Conformations and Binding Modes in Halogen-Bonded and Ionic Complexes of 2,3,5,6-Tetra(2'-pyridyl)pyrazine
- L5 ANSWER 5 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Zeolite catalyzed Waste-free Bromination of Aromatic Compounds
- L5 ANSWER 6 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Method to inhibit growth of microorganisms in aqueous systems and on substrates using a persulfate and a bromide
- L5 ANSWER 7 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Production method of bromo substitution azetidinone chemical compound
- L5 ANSWER 8 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Preparation of stabilized active bromine biocide solutions
- L5 ANSWER 9 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Control of Pseudomonas aeruginosa biofilms with **bromine**-based microbicides

- L5 ANSWER 10 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Effect of Temperature on Local Structure in Poly(ethylene oxide)-Zinc Bromide Salt Complexes
- L5 ANSWER 11 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Equilibrium and Kinetics Studies of Reactions of Manganese Acetate, Cobalt Acetate, and Bromide Salts in Acetic Acid Solutions
- L5 ANSWER 12 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Sulfamate stabilization of a bromine biocide in water
- L5 ANSWER 13 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI High-yield (meth)acrylate ester monomer preparation using alkali metal alkoxides as transesterification catalysts and bromide salts as polymerization inhibitors
- L5 ANSWER 14 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Carbonylation process and supported rhodium catalysts for the preparation of carboxylic acids from alkenes
- L5 ANSWER 15 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- Complexes of the (1R)-(+)-camphor azine diphosphines Z,Z-3,3'-Ph2PnC10H15=N-N=C10H15PxPh2 and Z,Z-3,3'-Ph2PxC10H15=N-N=C10H15PxPh2 (x=exo, n=endo) with Group 6 metal carbonyls: crystal structures of the ligands and fac-[W(CO)3(E,Z-Ph2PxC10H15=N-N=C10H15PxPh2)]
- L5 ANSWER 16 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Formation of Bromocarbenium Bromide Ion Pairs in the Electrophilic Bromination of Highly Reactive Olefins in Chlorinated Aprotic Solvents
- L5 ANSWER 17 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Continuous process for the manufacture of terephthalic acid of monomer purity by liquid-phase catalytic oxidation of p-xylene
- L5 ANSWER 18 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Continuous process for the manufacture of isophthalic acid of monomer purity by liquid-phase catalytic oxidation of m-xylene
- L5 ANSWER 19 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI The effect of pyridine ring functionalization on the conductivity of crosslinked CO-poly(styrene-4-vinylpyridine) derivatives
- L5 ANSWER 20 OF 37 CAPLUS COPYRIGHT 2006-ACS on STN
- TI Effect of LiClO4 and the nature of the verdazyl indicator on the heterolysis rate of benzylidene bromide in acetonitrile
- L5 ANSWER 21 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Face Selection in Additions to the Trigonal C2 Site in Quaternized 5-Azaadamantane Derivatives
- L5 ANSWER 22 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Charge Transfer Salts of Benzene-Bridged 1,2,3,5-Dithiadiazolyl Diradicals. Preparation, Structures, and Transport Properties of 1,3- and 1,4-[(S2N2C)C6H4(CN2S2)][X] (X = I, Br)
- L5 ANSWER 23 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Temperature gradient batteries
- L5 ANSWER 24 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI A chemical model for the Amoco "MC" oxygenation process to produce terephthalic acid

- L5 ANSWER 25 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Method for the preparation of 4-alkoxy-2-halomethyl-4-methyltetrahydrofurans
- L5 ANSWER 26 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Development of the zinc/bromine battery at Johnson Controls, Inc.
- L5 ANSWER 27 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI **Bromine** surface treatment of photosensitive elastomeric flexographic printing plates
- L5 ANSWER 28 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI **Bromine** surface treatment of photosensitive elastomeric flexographic printing plates
- L5 ANSWER 29 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Metal bromides
- L5 ANSWER 30 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Commensurate Peierls transition in a quasi-one-dimensional compound: the **bromide salt** of tetrathiafulvalene (TTF)
- L5 ANSWER 31 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Production of inorganic chlorides and bromides
- L5 ANSWER 32 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Bromine
- L5 ANSWER 33 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Bromine addition to chloronorbornene derivatives
- L5 ANSWER 34 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Bromine sanitization of waters
- L5 ANSWER 35 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Preparation of amines by catalytic reaction of terminal olefins with hydrogen bromide
- L5 ANSWER 36 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI Bimolecular displacement reactions. I. Reaction of endonorbornanol and 7-norbornanol with triphenylphosphine and **bromine**

DATE

- L5 ANSWER 37 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- TI The reaction of bromine with 9-acetylfluorene
- => d 1,29,31 bib ab
- L5 ANSWER 1 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
- AN 2006:36833 CAPLUS
- DN 144:110145
- TI Catalytic conversion of alkanes into liquid oxygenates
- IN Waycuilis, John J.
- PA USA
- SO U.S. Pat. Appl. Publ., 12 pp.
 - CODEN: USXXCO
- DT Patent
- LA English
- FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO.

----------A1 PΙ US 2006009662 20060112 US 2004-750984 20040102 PRAI US 2004-750984 20040102 A process is described for converting an alkane into an oxygenated product by passing an alkane gas over a first fixed bed containing a higher valence bromide salt to produce an alkyl bromide, a hydrobromic acid, and a lower valence bromide salt. The alkyl bromide and hydrobromic acid are conveyed as a gas to a second fixed bed containing a metal oxide and are passed over the second fixed bed to produce the first bromide salt and the oxygenated product. The metal oxide in the second fixed bed is regenerated by passing oxygen over the second fixed bed producing the metal oxide and bromine. The bromine is conveyed as a gas from the second fixed bed to the first fixed bed. The first bromide salt of the first fixed bed is regenerated by passing the bromine over the first fixed bed producing the first bromide salt. Process flow diagrams are presented. ANSWER 29 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN AN 1981:142066 CAPLUS DN 94:142066 ΤI Metal bromides IN Keblys, Kestutis A. PA Ethyl Corp., USA SO U.S., 4 pp. CODEN: USXXAM DTPatent LΑ English FAN.CNT 1 KIND DATE APPLICATION NO. PATENT NO. DATE ---------------_____ PI US 4248850 A 19810203 US 1979-15067 PRAI US 1979-15067 A 19790226 US 4248850 19790226 Metal bromides, especially alkali metal and alkaline earth bromides, are manufactured from the basic metal compound and Br2 in the presence of HCHO added as a reducing agent. Thus, a slurry of slaked lime 151.9 q in tap water 221.1 g was treated initially with a portion of a 37.4% formalin solution 80.1 g and then the Br2 (303.6 g) flow was started, both reagents being added simultaneously but the HCHO at a slightly faster rate. After filtration at 90° the product was 642 g of solution at pH 5.6. L5 ANSWER 31 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN ΑN 1980:474902 CAPLUS DN 93:74902 ΤI Production of inorganic chlorides and bromides ΑU Lebedev, O. V.; Artamonov, Yu. F. CS USSR SO Khimicheskaya Promyshlennost (Moscow, Russian Federation) (1980), (1), CODEN: KPRMAW; ISSN: 0023-110X DTJournal LΑ Russian AB An apparatus and method are proposed for the synthesis of highly concentrated orsaturated solns. of metal bromides and chlorides by the reaction of a metal salt or hydroxide with NH3 and Br2 or Cl2. The method was used for the

preparation of bromides and chlorides of NH4, Li, Na, K, Rb, Mg, Ca, Ba, Gd,

Cu, Zn, Pb, Y, Cd, Fe, and Al.

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